

High-pressure behavior of iron carbide (Fe₇C₃) at inner core conditions

Mainak Mookherjee,¹ Yoichi Nakajima,¹ Gerd Steinle-Neumann,¹ Konstantin Glazyrin,¹ Xiang Wu,^{1,2} Leonid Dubrovinsky,¹ Catherine McCammon,¹ and Aleksandr Chumakov³

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[1] Carbon is a plausible light element candidate in the Earth's core owing to its cosmic abundance and its chemical affinity for iron. Recent experimental studies on Fe-C phase relations at high pressures have demonstrated that Fe₇C₃ iron carbide is a likely candidate for the Earth's inner core. Using electronic structure calculations, we determine the equation of state, the full elastic constant tensor, and the sound wave velocities for Fe₇C₃, up to inner core pressures. We find that Fe₇C₃ is ferromagnetic (*fm*) at low pressure, and that its compression behavior is well represented by a third-order Birch Murnaghan finite strain expression with $V_0^{fm} = 9.1 \text{ \AA}^3/\text{atom}$, $K_0^{fm} = 231 \text{ GPa}$, and $K_0^{\prime fm} = 4.4$. Under compression the magnetic moments of the Fe atoms gradually decrease, and at ~67 GPa the magnetic moment is lost. The high-pressure nonmagnetic phase (*nm*) has distinct finite strain parameters with $V_0^{nm} = 8.8 \text{ \AA}^3/\text{atom}$, $K_0^{nm} = 291 \text{ GPa}$, and $K_0^{\prime nm} = 4.5$. Calculated elastic constants show softening associated with the loss of magnetization. In addition, we have conducted nuclear resonant inelastic X-ray scattering experiments on ⁵⁷Fe enriched Fe₇C₃ at 1 bar and 300 K. On the basis of our nuclear resonant inelastic X-ray scattering spectra we have derived a Debye sound velocity of 3.18 km/s. The experimentally determined value is in good agreement with the computational predictions, based on athermal single elastic constants. The static *P* wave velocity at inner core pressures agrees well with seismological constraints, whereas the *S* wave velocity is greater by 30%. On the basis of the density of Fe₇C₃ at inner core conditions, we predict that the maximum possible carbon content of the inner core is around 1.5 wt %.

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1. Introduction

[2] The Earth's core is significantly denser than the overlying mantle and consists of a metallic liquid outer core and a coexisting solid inner core that has been freezing out of the outer core through Earth's history [Jacobs, 1953]. From cosmochemical and geochemical arguments, there is a consensus in the Earth and planetary sciences community that the core is composed primarily of iron (Fe). Comparison of seismological data with experimental measurements on the compressibility of Fe shows that the Earth's outer core is lighter than pure iron by 5–10% and the inner core is lighter than solid iron by 3–7% (the so-called density deficit of the core) [Birch, 1964; Brown and McQueen, 1986; Jephcoat and Olson, 1987]. The density deficit of the core requires

the presence of lighter elements [Stevenson, 1981], the nature of which has been the subject of extensive reviews in geochemistry [McDonough, 2003; Li and Fei, 2003] and geophysics [Stixrude et al., 1997; Badro et al., 2007] where many light elements have been identified as potential candidates and trade-offs are discussed. In exploring the various possibilities, it is to be noted that the light element should readily alloy with iron and be cosmochemically abundant. With these constraints, S, O, Si, C and H have been the most preferred candidates [Hillgren et al., 2000].

[3] Distribution of carbon in various reservoirs of the Earth has been extensively researched, and there is little doubt that the solid Earth is the most important reservoir [Javoy et al., 1982; Sleep and Zahnle, 2001]. However, the relative distribution between the metallic core and the silicate mantle is poorly understood.

[4] Some models of the carbon budget for the solid Earth [Sleep and Zahnle, 2001; Coltice et al., 2004] do not include carbon in the Earth's core. On the other extreme, Kuramoto [1997] proposed complete partitioning of carbon to the core, based on thermodynamic models of gas solubilities in silicate and metallic melts. Geochemical arguments suggest that

¹Bayerisches Geoinstitut, Universität Bayreuth, Bayreuth, Germany.

²School of Earth and Space Sciences, Peking University, Beijing, China.

³ID18 Nuclear Resonance Group, European Synchrotron Radiation Facility, Grenoble Cedex, France.

carbon is one of the light alloying elements [Wood, 1993; Poirier, 1994; Li and Fei, 2003; Huang et al., 2005] with concentration as high as 4 wt % [Hillgren et al., 2000] and as low as 0.2 wt % [McDonough, 2003]. If carbon were the dominant light element in the core, then the core would have an orders of magnitude higher carbon budget than the mantle. If carbon were a minor light element, the contribution of the core to the total terrestrial carbon budget would be comparable to or smaller than that of the mantle [Dasgupta and Walker, 2008; Dasgupta and Hirschmann, 2010].

[5] Cosmochemically, less than 3×10^{-4} wt % of C is expected to be present in Fe condensed from the solar nebula, owing to a large difference in the condensation temperatures of C (<600 K) and Fe (~1470 K) at 10^{-3} atm [Morgan and Anders, 1980; Anderson, 2007]. In addition, C might have been partially lost owing to high-temperature processes during or after accretion [Hillgren et al., 2000].

[6] Nevertheless, carbon is abundant with 3.6 wt % in carbonaceous chondrites (CI) [Anderson, 2007], and at the slightly elevated pressures (0.01–5.00 GPa) prevalent during planetary accretion and differentiation, the solubility of C in Fe is significantly enhanced based on the persistence of a eutectic point in the Fe-C systems up to 12 GPa as determined experimentally [Hirayama et al., 1993] and in thermodynamic calculations [Wood, 1993]. The importance of such high-pressure processes is highlighted by the widespread occurrences of graphite and carbides in iron meteorites [Wasson, 1985]. Slow cooling rates on the order $0.4\text{--}4^\circ\text{C Ma}^{-1}$ in many of these meteorites as determined by Widmanstätten textures [Goldstein and Short, 1967b, 1967a] indicate that the graphite and carbide-bearing iron meteorites are fragments of cores from planetesimals with radii of at least 100–400 km. Iron-based carbides, in the form of cohenite (Fe,Ni)₃C, which is similar in structure to cementite Fe₃C, have been observed in such meteorites [Brett, 1966; Wasson, 1985].

[7] On the basis of the extrapolation of thermodynamic data to inner core conditions, Wood [1993] proposed Fe₃C (6.7 wt % carbon) as an ideal inner core candidate, coexisting with pure Fe, as also observed in recent experiments [Tateno et al., 2010]. This notion found support in ab initio computations that predicted the stability of coexisting Fe and Fe₃C over C dissolved in Fe [Huang et al., 2005]. However, the proposed presence of Fe₃C has been questioned from two different angles: the elastic properties of Fe₃C at pressures of the Earth's core [Vočadlo et al., 2002], and phase equilibria studies in the Fe-C binary system at pressures above 130 GPa [Lord et al., 2009].

[8] On the basis of systematics of compressibility in intermetallic phases, Wood [1993] assumed a bulk modulus $K_0 = 174$ GPa and its pressure derivative $K'_0 = 5.1$ for Fe₃C. X-ray diffraction studies on the equation of state of Fe₃C up to 73 GPa [Scott et al., 2001] and 30 GPa [Li et al., 2002] were in agreement with the parameters used by Wood [1993], but subsequent ab initio simulations by Vočadlo et al. [2002] revealed a magnetic transition (high-spin low-spin transition) at ~60 GPa that had a strong effect on the thermoelastic properties of Fe₃C at core pressures. In this study the high-pressure nonmagnetic phase was found to be substantially stiffer ($K_0 = 316$ GPa) than the previously measured bulk moduli [Scott et al., 2001; Li et al., 2002].

The experiments were restricted to conditions below or near the predicted transition pressures and were insensitive to magnetism. The choice of a low-bulk modulus by Wood [1993] led to an inaccurate extrapolation of phase equilibria at pressures relevant to inner core.

[9] In addition, the Fe-C phase diagram has been revisited and extended to higher pressures. On the basis of density measurements of liquid Fe₃C up to 9.5 GPa, Terasaki et al. [2010] concluded that carbon is likely to be partitioned into the solid phase, which for the Earth implies the inner core. Nakajima et al. [2009] conducted multianvil quench experiments to 14 GPa and in situ X-ray diffraction up to 29 GPa. They observed a new phase with Fe₇C₃ stoichiometry as a stable liquidus phase above 10 GPa, in contrast to the previously assumed Fe₃C cementite phase. Lord et al. [2009] recently extended the pressure range for phase equilibria studies in the Fe-C system to 70 GPa using the laser heated diamond anvil cell technique, and concluded that Fe₇C₃ is the liquidus phase up to core pressures. The occurrence of Fe₇C₃ as the liquidus phase in the ternary system Fe-C-S [Dasgupta et al., 2009] further strengthens the importance of this phase for the core of the Earth and other planets.

[10] With Fe₇C₃ as a potential inner core phase, its thermoelastic parameters at inner core pressures are of geophysical and geochemical interest. In this article we determine the full elastic constant tensor of Fe₇C₃ at high pressure from first principles electronic structure simulations. Similar to the behavior of Fe₃C under compression, the possibility of a high-spin low-spin transition can have a significant effect on elastic parameters of Fe₇C₃. Hence we examine the behavior of both the magnetic and nonmagnetic phases of Fe₇C₃ at high pressure.

2. Method

2.1. Electronic Structure Method

[11] The electronic structure and energetics of the iron carbide Fe₇C₃ have been calculated with a density functional theory (DFT) based method [Kohn and Sham, 1965]. DFT describes the electronic structure of materials through the electron density, translating the solution for the many body wave function and associated energies to a number of one electron problems in the Kohn-Sham equations [Kohn and Sham, 1965]. The DFT solution for the energies in the system considered is exact in principle, but the nature of the electronic many-body interaction is mapped onto a potential for exchange and correlation that needs to be approximated. Here we have used the generalized gradient approximation (GGA-PBE) [Perdew et al., 1996] that has been shown to yield good results for transition metals [Steinle-Neumann et al., 1999; Alfè et al., 2002b; Steinle-Neumann et al., 2004; Steinle-Neumann, 2008] and some of its alloys [Skorodumova et al., 2004; Isaev et al., 2007; Cote et al., 2008], including iron carbides [Vočadlo et al., 2002; Huang et al., 2005]. The electronic structure of the Fe-carbides is different from transition metal oxides and silicates [Cococcioni and de Gironcoli, 2005; Alfredsson et al., 2004] for which standard approximations to exchange and correlation fail because of very localized electronic states, and theory beyond DFT has to be applied to accurately describe their properties [Kolorenč and Mitas, 2008];

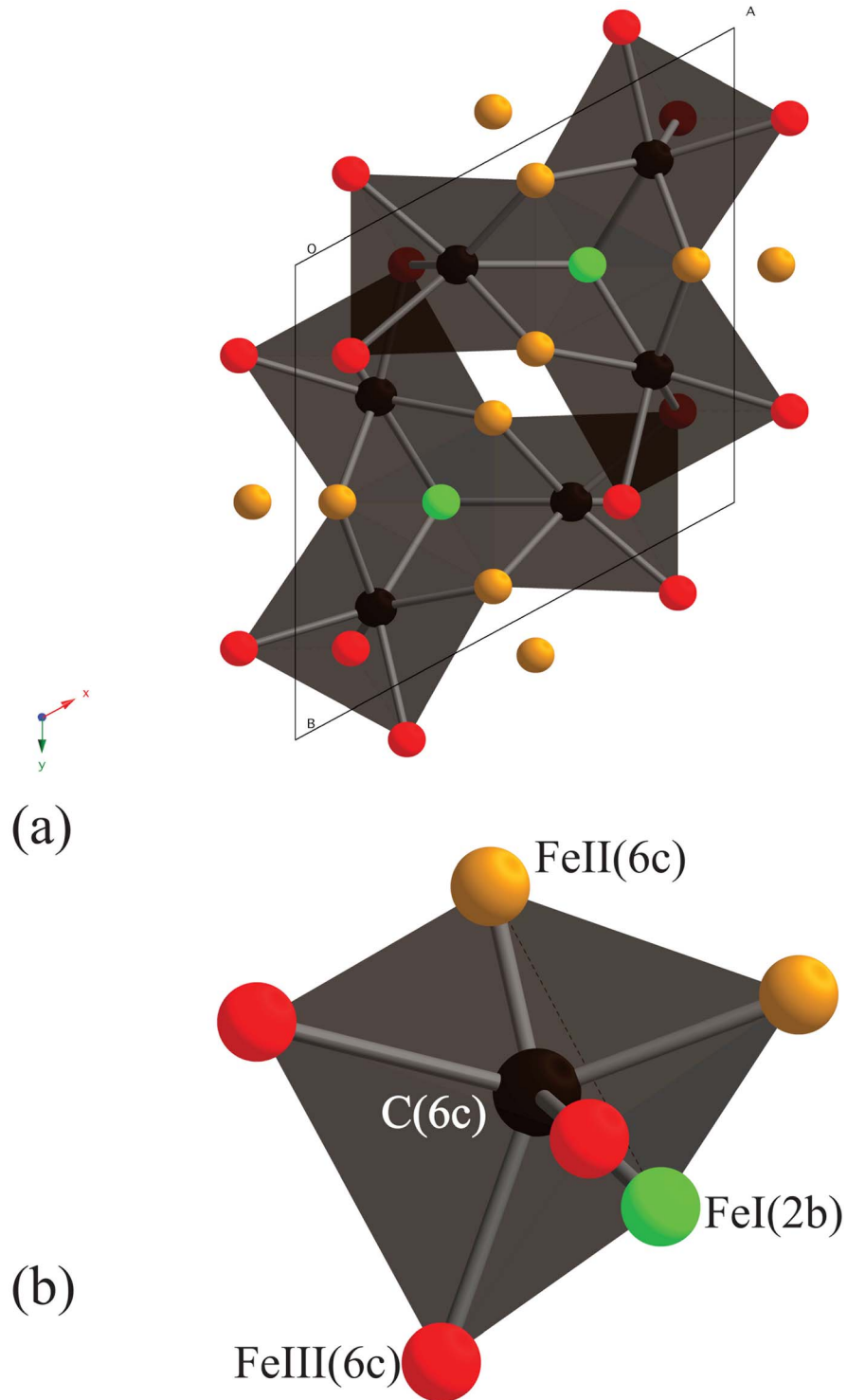


Figure 1. (a) Crystal structure of Fe_7C_3 with hexagonal symmetry, $P6_3mc$. (b) The iron atoms form a ditrigonal prism around the central carbon atom (black). Three such prisms are combined in a triangular arrangement. There are three distinct Fe sites: FeI (Wyckoff site 2b, green), FeII (Wyckoff site 6c, golden), and FeIII (Wyckoff site 6c, red). There is only one site for the carbon atom, C (Wyckoff site 6c). Each ditrigonal prism has three FeIII atoms, two FeII atoms, and one FeI atom. The FeI atom is located at the center of the triangle as shown in Figure 1a.

Kuneš *et al.*, 2009]. Here we use the highly accurate projector augmented wave method (PAW) [Kresse and Joubert, 1999] as implemented in the Vienna ab initio simulation package (VASP) [Kresse and Hafner, 1993; Kresse and Furthmüller, 1996] which uses a plane wave basis set. We include plane waves up to cutoff energy (E_{cut}) of 1000 eV for the expansion of the charge density and the GGA-PBE atomic files provided within the VASP package [Kresse and Joubert, 1999].

[12] The crystal structure of the iron carbide Fe₇C₃ (Eckstrom-Adcock carbide) has hexagonal $P6_3mc$ space group (186) [Herbstein and Snyman, 1964] and contains 20 atoms in the primitive unit cell (Figure 1). The number of reciprocal space vectors at which the Kohn-Sham equations are solved (k -points) was chosen as $4 \times 4 \times 4$ [Monkhorst and Pack, 1976], which translates to 12 k -points within the irreducible wedge of the Brillouin zone. In addition to relaxing the magnetic moment in spin-polarized calculations for the ferromagnetic (fm) structure and two ferrimagnetic arrangements of Fe₇C₃ (with an antiferromagnetic nature), nonmagnetic (nm) calculations were performed for the whole compression range to study the influence of magnetism on the elastic parameters. The ferrimagnetic phases were created with spin up on the FeI (Wyckoff site 2b, Figure 1) and FeIII (Wyckoff site 6c) sites and spin down on the FeII (Wyckoff site 6c) site ($afm1$), as well as spin up on the FeI and FeII sites and spin down on the FeIII sites ($afm2$).

[13] In order to determine the relative stability of Fe₇C₃ and Fe₃C we have also calculated the energetics of Fe₃C, Fe and C. Fe₃C has an orthorhombic space group $Pnma$ with $Z = 4$ [Fasiska and Jeffrey, 1965; Vočadlo *et al.*, 2002]. The k -point mesh was chosen as $8 \times 8 \times 8$ [Monkhorst and Pack, 1976], resulting in 64 k -points within the irreducible wedge of the Brillouin zone. Hexagonal close packed (hcp) iron with space group $P6_3/mmc$ ($Z = 2$) is believed to be stable at inner core conditions [Mao *et al.*, 1990; Hemley and Mao, 2001; Nguyen and Holmes, 2004]. Above 0.7 GPa carbon adopts the diamond structure [Bundy, 1989] with space group $Fd\bar{3}m$ ($Z = 8$). Computations were performed for a $20 \times 20 \times 20$ k -point mesh for both hcp iron and diamond with 1100 and 220 k -points, respectively, within the irreducible wedge of the Brillouin zone. With the k -point and E_{cut} chosen, total energies were converged to within 1 meV/atom and stresses to within 0.1 GPa.

[14] Athermal elastic constants for Fe₇C₃ were determined for the fm and nm phases. Elastic constants were computed through the changes in the stress tensor ($\underline{\sigma}$) with respect to applied strain ($\underline{\epsilon}$) [Mookherjee and Steinle-Neumann, 2009a, 2009b]. We first calculated the equilibrium and relaxed structure at a given volume, V . Then we strained the lattice and let the internal degree of freedom of the crystal structure relax consistent with the symmetry. We applied positive and negative strains of magnitude $\delta = 1\%$ in order to accurately determine the stress in the appropriate limit of zero strain. The strained lattice \vec{a}' is related to the unstrained lattice \vec{a} by $\vec{a}' = (\underline{I} + \underline{\epsilon})\vec{a}$, where \underline{I} is the identity matrix. Four distinct strain tensors ϵ were applied: $\epsilon_{11} = \delta$; $\epsilon_{21} = \epsilon_{12} = \delta$; $\epsilon_{31} = \epsilon_{13} = \delta$; $\epsilon_{33} = \delta$ (all other entries of $\epsilon_{ij} = 0$) on the hexagonal lattice. These resulted in five independent elastic constants, c_{11} ($= c_{22}$), c_{12} , c_{13} ($= c_{23}$), c_{33} , c_{44} ($= c_{55}$), with $c_{66} = \frac{1}{2}(c_{11} - c_{12})$ [Nye, 1985].

2.2. Nuclear Resonant X-Ray Inelastic Scattering

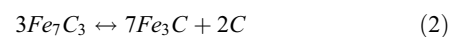
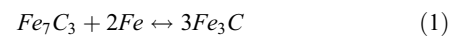
[15] Fe₇C₃ samples were synthesized from a powder mixture of iron and graphite with a atomic ratio of Fe:C = 2:1, with higher carbon content compared to the target stoichiometry. To obtain sufficient intensity in the phonon energy spectrum of the sample, iron was doped with 25 wt % of ⁵⁷Fe. The mixture was packed into an MgO capsule and equilibrated at 18 GPa and 1873 K for 3 h using a multianvil apparatus at Bayerisches Geoinstitut. The recovered sample consists of large homogeneous domains ($>100 \mu\text{m}$) of Fe₇C₃ as determined by scanning electron microprobe analysis with energy dispersive spectroscopy. The excess carbon was observed as diamond only at the surface of the MgO capsule. We carefully selected the large domains under the microscope for further experiments. Subsequent X-ray diffraction confirmed the Fe₇C₃ phase with hexagonal symmetry ($a = 6.88 \text{ \AA}$ and $c = 4.53 \text{ \AA}$). Minor peaks of MgO were observed.

[16] Nuclear resonant inelastic X-ray scattering (NRXIS) experiments were performed at the ID18 beamline of the European Synchrotron Radiation Facility (ESRF), Grenoble, France. The focused X-ray beam was less than 10 mm in diameter and has an energy resolution of 1.0 meV. Two avalanche photodiode detectors (APD) were used to collect the inelastic scattering in directions perpendicular to the X-ray beam. The diameter of the sample was $\sim 120 \mu\text{m}$. The NRXIS spectra were collected over a range of -80 to 80 meV around the ⁵⁷Fe nuclear resonance energy of 14.4 keV in steps of 0.2 meV. The collection time was 4 h.

3. Results

3.1. Energetics

[17] In order to assess the relative stability of Fe₇C₃ and Fe₃C, we have considered the following reactions:



Reactions (1) and (2) have different bulk composition with 6.7 and 8.4 wt % C, respectively. At pressures corresponding to inner core conditions, the Fe₇C₃, Fe₃C and Fe phases are nonmagnetic (see also section 3.2), and hence we have estimated the enthalpy based on the nonmagnetic energies at static conditions. For Fe we have considered the hcp phase as outlined above. At static conditions ($T = 0$ K) the free energy change associated with the reaction, $\Delta G = \Delta H - T\Delta S$, is equivalent to the enthalpy change, ($\Delta G = \Delta H$). For reaction (1) with 6.7 wt % C, we find that Fe₃C is energetically stable compared to a mixture of Fe and Fe₇C₃ at all pressures (Figure 2). This observation does not agree with the recent experimental findings conducted in the iron-carbon systems [Dasgupta *et al.*, 2009; Lord *et al.*, 2009; Nakajima *et al.*, 2009] where Fe₇C₃ is a stable liquidus phase at higher pressures. This discrepancy can be traced to the considerably stiffer bulk modulus computed for hcp iron compared to experiments (Table 1). The discrepancy between measured and computed values of K_0 is consistent with previous ab initio studies [Vočadlo *et al.*, 1997; Steinle-Neumann *et al.*, 1999].

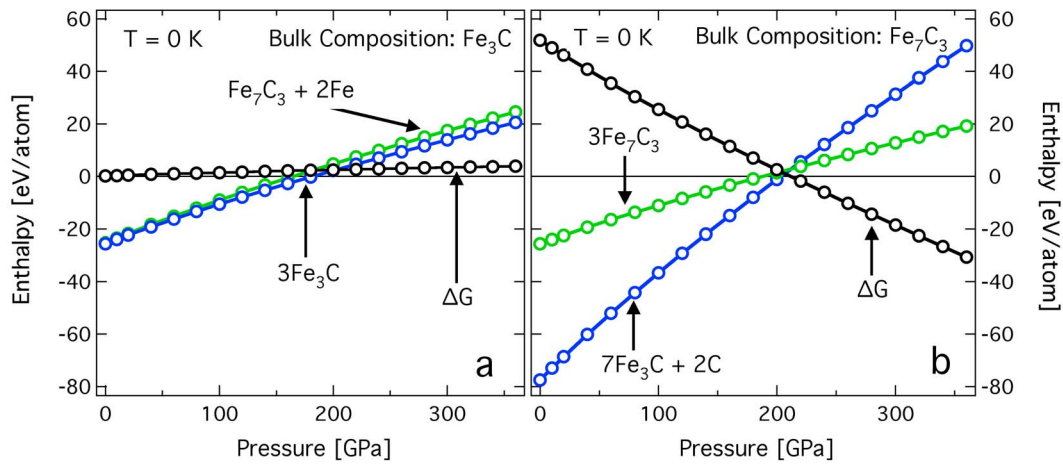


Figure 2. Energetics of the reaction (a) $\text{Fe}_7\text{C}_3 + 2\text{Fe} \leftrightarrow 3\text{Fe}_3\text{C}$ and (b) $3\text{Fe}_7\text{C}_3 \leftrightarrow 7\text{Fe}_3\text{C} + 2\text{C}$ at static conditions, $T = 0$ K. All calculations are for nonmagnetic phases. Note that for the bulk carbon content of 6.7 wt %, i.e., Figure 2a, the enthalpy of Fe_3C is lower than the mixture of Fe_7C_3 and Fe-hcp phase. With a bulk carbon content of 8.4 wt %, i.e., Figure 2b, the Fe_7C_3 phase is stabilized over Fe_3C and C-diamond at inner core conditions.

[18] For reaction (2) with bulk 8.4 wt % C, we find that Fe_3C and C (diamond) is the stable assemblage at lower pressure; however, above 200 GPa Fe_7C_3 is stabilized over Fe_3C and C. It is likely that temperature or other elements

present in the core might affect the relative stability of Fe_7C_3 versus Fe_3C . The role of temperature could be examined by molecular dynamics simulations or by estimating the vibrational entropy through lattice dynamics. However, both

Table 1. Equation of State Parameters and Average Magnetic Moment (M) per Fe Atom in Atomic Units (μ_B), for Iron Carbides, Hexagonal Closed Packed (hcp) Iron, and Diamond^a

| Space Group | Phase | E_0 (eV/atom) | V_0 (cm ³ /mol) | K_0 (GPa) | K' | M (μ_B) | Reference |
|------------------------------------|-------|-----------------|------------------------------|-------------|------|-----------------|--|
| <i>Fe₇C₃</i> | | | | | | | |
| <i>P6₃mc</i> | nm | -8.53 | 52.82 | 291 | 4.5 | | this work (PBE-PAW) |
| | | -8.54 | 52.35 | 339 | 4.0* | | |
| | fm | -8.60 | 54.81 | 231 | 4.4 | 1.63 | this work (PBE-PAW) |
| | | -8.60 | 54.79 | 238 | 4.0 | 1.63 | |
| | | -8.58 | 54.28 | 233 | 5.0 | 0.82 | |
| | afm1 | -8.58 | 54.41 | 243 | 4.0* | 0.82 | this work (PBE-PAW) |
| | | -8.57 | 53.99 | 223 | 6.0 | 0.04 | |
| afm2 | -8.57 | 54.22 | 242 | 4.0* | 0.04 | | |
| exp. | | | 55.47 | 253 | 3.6 | | Nakajima et al. (submitted manuscript, 2010) |
| <i>Fe₃C</i> | | | | | | | |
| <i>Pnma</i> | nm | -8.50 | 21.58 | 298 | 4.6 | | this work (PBE-PAW) |
| | | -8.51 | 21.54 | 329 | 4.0* | | |
| | | | 21.61 | 317 | 4.3 | | |
| | fm | -8.59 | 22.92 | 223 | 3.1 | 1.92 | this work (PBE-PAW) |
| | | -8.59 | 22.93 | 211 | 4.0* | 1.92 | |
| exp. | | | 23.08 | 229 | 5.4 | 1.88 | <i>Vocadlo et al. [2002]</i> <i>Scott et al. [2001]</i> <i>Li et al. [2002]</i> |
| <i>Fe (hcp)</i> | | | | | | | |
| <i>P6₃mmc</i> | nm | -8.37 | 6.17 | 296 | 4.4 | | this work (PBE-PAW) |
| | | -8.37 | 6.12 | 296 | 4.0* | | |
| | | | 5.90 | 292 | 4.4 | | |
| | exp. | | 6.26 | 290 | 4.0 | | <i>Steinle-Neumann et al. [1999]</i> <i>Vocadlo et al. [1997]</i> <i>Mao et al. [1990]</i> <i>Dewaele et al. [2006]</i> |
| | | | 6.73 | 165 | 5.3 | | |
| | | 6.75 | 165 | 5.0 | | | |
| <i>C (Diamond)</i> | | | | | | | |
| <i>Fd3m</i> | comp. | -9.06 | 3.44 | 434 | 3.6 | | this work (PBE-PAW) |
| | | -9.06 | 3.45 | 377 | 4.0* | | |
| | exp. | | 3.41 | 446 | 3.0 | | <i>Ocellli et al. [2003]</i> |

^a E_0 , V_0 , K_0 , and K' are zero pressure energy, volume, bulk modulus, and its pressure derivative, respectively. All computations were fit with a third- and second-order ($K' = 4.0$, indicated by asterisk for fixed K') Eulerian finite strain equation of state.

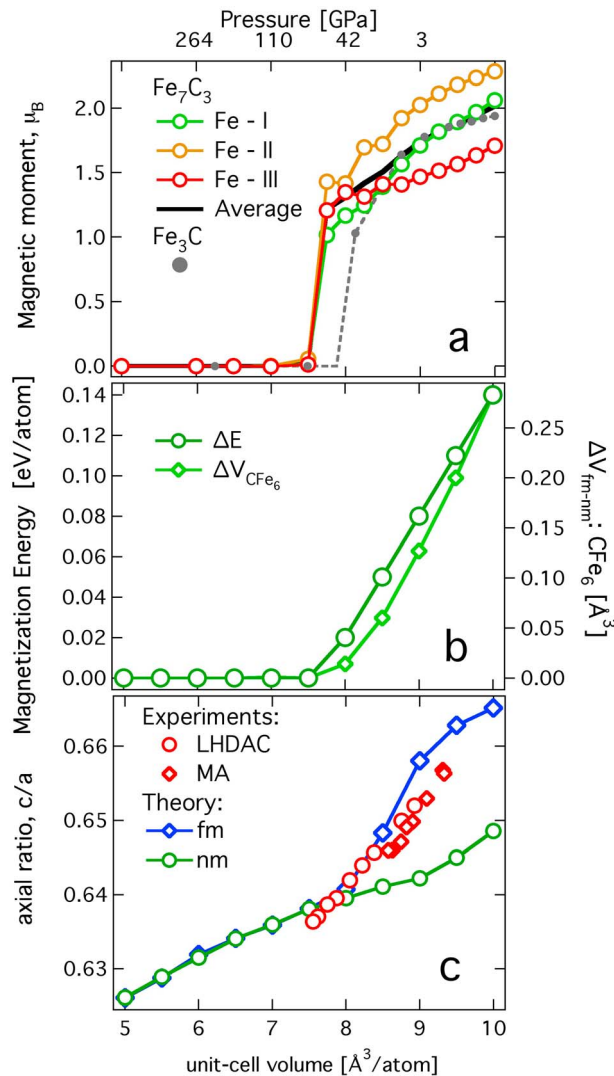


Figure 3. (a) Variation of magnetic moments for the ferromagnetic phase of Fe₇C₃ as a function of volume. Shown are values for FeI (green), FeII (golden), and FeIII (red). The black line shows the average magnetic moment, and the gray filled circles and dashed line show the average magnetic moment of Fe₃C [Vočadlo *et al.*, 2002]. The magnetic moments collapse at 7.5 Å³/atom for Fe₇C₃. (b) Magnetization energy, i.e., the energetic difference between the magnetic and nonmagnetic phases at the same volume, for Fe₇C₃ also vanishes at 7.5 Å³/atom. The volume difference for the ditrigonal CFe₆ polyhedral units between the *fm* and *nm* phases are also shown in Figure 3b. (c) The *c/a* ratio for Fe₇C₃ for the ferromagnetic (*fm*, blue open rhomb and line) and nonmagnetic structures (*nm*, green open circle and line). Red open rhomb (laser heated diamond anvil cell) and open (multianvil, 300 K) circle are experimental data (Nakajima *et al.*, submitted manuscript, 2010).

methods are rather expensive given the large and complex system in question, and it is beyond the scope of the current study.

[19] Nevertheless, with the experimental findings and the theoretical confirmation (for 8.4 wt % C bulk composition)

that Fe₇C₃ is stable at high pressures and forms the liquidus phase [Dasgupta *et al.*, 2009; Lord *et al.*, 2009; Nakajima *et al.*, 2009], it is clearly worth exploring its physical properties using DFT-based computations.

3.2. Magnetism

[20] Ferromagnetic Fe₇C₃ is found to be energetically stable over both ferrimagnetic configurations (intermediate in energy) and over the nonmagnetic solution for all volumes larger than 7.5 Å³/atom, where the magnetic moment is lost from the structure on all Fe sites and the magnetization energy drops to zero (Figure 3). The magnetic moment per Fe for the ferromagnetic phase is 1.63 μ_B at V₀^{fm} ~ 9.1 Å³/atom (Table 1), with individual moments being 1.71 μ_B for FeI, 2.03 μ_B for FeII, and 1.47 μ_B for FeIII. For the *afm1* and *afm2* phases the net magnetic moments are 0.82 and 0.04 at V₀^{afm1} and V₀^{afm2}, respectively. They consist of 2.05 (1.21) μ_B for FeI, -0.42 (1.19) μ_B for FeII and 1.66 (-1.52) μ_B for FeIII in the *afm1* (*afm2*) structures. In comparison, the average magnetic moment for Fe₃C based on theory and experiment at V₀^{fm} is 1.88 μ_B [Vočadlo *et al.*, 2002] and 1.72–1.78 μ_B [Hofer and Cohn, 1959; Shabanova and Trapeznikov, 1973].

[21] For Fe₇C₃, Tsuzuki *et al.* [1984] measured the magnetic moment to be 1.31 μ_B at ambient conditions, which is significantly lower than our calculations. A similar value of 1.35 μ_B for Fe₃C was reported in their study, which is also lower than that in previous studies. As pointed out by Tajima and Hirano [1990], lower values by Tsuzuki *et al.* [1984] may be caused by the imperfection in sample preparation. It is therefore necessary to reexamine the magnetic moment of Fe₇C₃.

[22] The pressure of magnetic collapse (67 GPa) is very similar to that of cementite Fe₃C predicted from DFT-based computations (~60 GPa) [Vočadlo *et al.*, 2002; Ono and Mibe, 2010]. This transition pressure corresponds to anomalous behavior in inelastic X-ray scattering dispersion above 68 GPa [Fiquet *et al.*, 2009]. However, high-pressure magnetic measurements indicate a lower transition pressure, although there is considerable discrepancy between different studies. A synchrotron Mössbauer spectroscopy study on Fe₃C implies a magnetic collapse between 4.3 and 6.5 GPa [Gao *et al.*, 2008]. X-ray circular dichroism measures loss of magnetism near 10 GPa [Duman *et al.*, 2005], and by X-ray emission spectroscopy the transition pressure is determined to be approximately 25 GPa [Lin *et al.*, 2004].

3.3. Equation of State and Structure Under Compression

[23] As mentioned above, the magnetization energy (ΔE_{nm-fm}) decreases linearly and vanishes at 7.5 Å³/atom, i.e., at 67 GPa, in agreement with the magnetic collapse (Figure 3).

[24] In the volume range 10.5 to 7.5 Å³/atom, i.e., the region where the *fm* solution is energetically favored, the *c/a* ratio in the *fm* phase is larger than that of the *nm* phase (Figure 3). As the structure is compressed, the *c/a* ratio for the *fm* phase is reduced and it approaches the *c/a* ratio for the *nm* phase. They become equal at 7.5 Å³/atom, corresponding to the magnetic collapse (Figure 3). The predicted decrease in *c/a* for the *fm* phase is in good agreement with experimental measurements (Y. Nakajima *et al.*, Thermo-

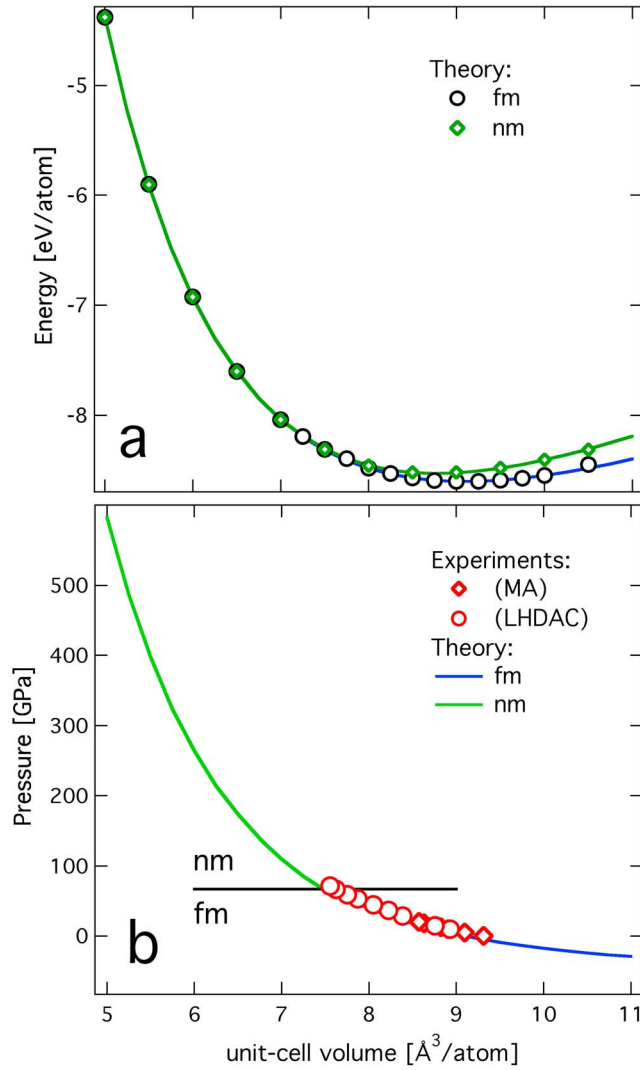


Figure 4. Energetics and equation of state of Fe₇C₃. (a) Plot of computed energy as a function of volume for the ferromagnetic (*fm*, black open circles) and nonmagnetic (*nm*, green open rhombs) phase. A finite strain fit to the computed *fm* energies between 10.5 and 7.75 Å³/atom is shown by the blue line. The green line shows the finite strain fit to the *nm* energies. (b) Blue and green lines represent the computed P-V Birch Murnaghan equations of state for Fe₇C₃ at low and high pressure, respectively. Red symbols show data from laser heated diamond anvil cell experiments (open circles) and multi-anvil experiments at 300 K (open rhombs) from the work of Nakajima et al. (submitted manuscript, 2010).

elastic properties of Fe₇C₃: Implication for carbon in the Earth's inner core, submitted to *American Mineralogist*, 2010). As is the case for elemental cobalt [Antonangeli et al., 2008], the *c/a* ratio appears to be a very sensitive indicator for the presence of magnetism at low pressure and its loss at high pressure.

[25] At 10 Å³/atom the volume difference of the ditrigonal prism CFe₆ (Figure 1) between the *fm* and *nm* phases is around 0.3 Å³. The volume differences decrease linearly

upon compression and mimic the behavior of the magnetization energy (Figure 3).

[26] The calculated energy-volume relations for *fm* and *nm* Fe₇C₃ are shown in Figure 4. For the *fm* phase we find that a third-order Birch-Murnaghan finite strain equation of state (BM-EoS) [Birch, 1947] adequately describes energy-volume results between 10.5 and 7.75 Å³/atom, with $V_0^{fm} = 9.1$ Å³/atom, $K_0^{fm} = 231$ GPa and $K_0^{\prime fm} = 4.4$. The energy-volume relationship for the *nm* phase between 10.5 and 5.0 Å³/atom is described by BM-EoS parameters of $V_0^{nm} = 8.77$ Å³/atom, $K_0^{nm} = 291$ GPa, and $K_0^{\prime nm} = 4.5$ (Table 1). Similar to Fe₃C, there is a significant difference in the bulk moduli of the phases, with the *nm* Fe₇C₃ being considerably stiffer than the *fm* phase, as illustrated by the energy difference between the *nm* and *fm* E-V EoS (Figure 4).

3.4. Elasticity

[27] For both the *fm* and *nm* phases, the elastic constants increase upon compression. At 67 GPa, where magnetism collapses, an anomalous reduction for some of the elastic constants for the *fm* phase (c_{11}^{fm} , c_{12}^{fm} and c_{44}^{fm} , in Voigt notation c_{ij}) is predicted (Figure 5). Upon further compression, all the elastic constants increase monotonically. The elastic constants of the *nm* phase increase monotonically with pressure, merging with those of the *fm* phase after the loss of magnetism. Finite strain fits to the elasticity data (now in full fourth-rank tensor notation c_{ijkl}) were made using the formulation

$$c_{ijkl} = (1 + 2f)^{3.5} [c_{ijkl0} + b_1 f + 0.5b_2 f^2] - P\Delta_{ijkl}, \quad (3)$$

where

$$f = 0.5 \left[(V_0/V)^{2/3} - 1 \right] \quad (4)$$

is the finite strain and

$$b_1 = 3K_0(c'_{ijkl0} + \Delta_{ijkl}) - 7c_{ijkl0}, \quad (5)$$

$$b_2 = 9K_0^2 c''_{ijkl0} + 3K_0'(b_1 + 7c_{ijkl0}) - 16b_1 - 49c_{ijkl0}, \quad (6)$$

$$\Delta_{ijkl} = -\delta_{ij}\delta_{kl} - \delta_{ik}\delta_{jl} - \delta_{il}\delta_{jk}. \quad (7)$$

c'_{ijkl0} and c''_{ijkl0} are the first and second derivatives of c_{ijkl} with respect to pressure. Δ_{ijkl} takes a value of -3 for longitudinal (c_{iiii} in full tensor and c_{ii} in Voigt notation, with $i = 1, 2, 3$) and off-diagonal elastic constants (c_{ijij} in full tensor and c_{ij} in Voigt notation, with $i = 1, 2, 3, i \neq j$), -1 for shear constants (c_{ijij} in full tensor notation with $i = 1, 2, 3, i \neq j$ and c_{ij} in Voigt notation with $i = 4, 5, 6, i = j$), and 0 otherwise [Stixrude and Lithgow-Bertelloni, 2005]. At pressures below the magnetic collapse, $c_{11}^{nm} > c_{11}^{fm}$ and $c_{12}^{nm} > c_{12}^{fm}$, indicating that magnetism affects the [100] (and symmetrically equivalent [010]) direction in the Fe₇C₃ structure, making them softer than the values in the corresponding nonmagnetic structure. For the shear elastic constant $c_{44}^{fm} > c_{44}^{nm}$, i.e., the relationship between the *fm* and *nm* phases is opposite, indicating that aligned spins resist shearing. The dependent shear elastic constant, $c_{66}^{fm} =$

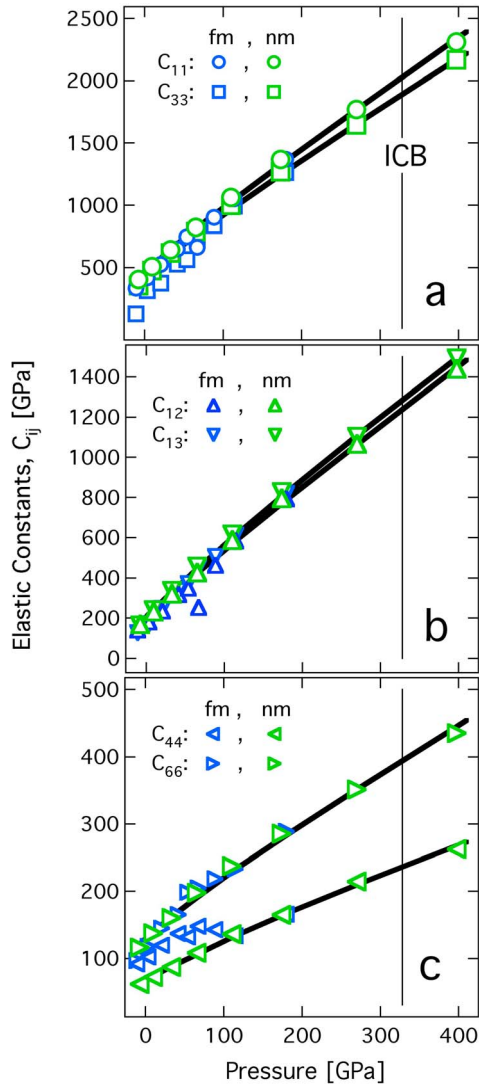


Figure 5. Single crystal elastic constants of Fe₇C₃ *fm* and *nm* phases as a function of pressure. (a) Longitudinal elastic constants: c_{11} (open circles) and c_{33} (open square); (b) off-diagonal elastic constants: c_{12} (upward pointing triangle) and c_{13} (downward pointing triangle). (c) Shear elastic constants: c_{44} (triangle pointing toward left) and c_{66} (triangle pointing toward right). Green symbols show the *nm* phase and blue symbols represent the *fm* phase. Lines are finite strain fits. Note that the *fm* elastic constants are lower than the *nm* elastic constants at low pressure except for c_{44} and c_{66} . Also, the elastic constants c_{11} and c_{12} of the magnetic phase exhibit softening at 67 GPa.

$0.5(c_{11}^{fm} - c_{12}^{fm})$, does not show this effect prominently and increases monotonically at all pressures.

[28] The isotropic bulk (K) and shear (G) moduli are determined using the relations

$$K_{Voigt} = (1/9)[2c_{11} + c_{33} + 2(c_{12} + 2c_{23})], \quad (8)$$

$$K_{Reuss} = [2s_{11} + s_{33} + 2(s_{12} + 2s_{23})]^{-1}, \quad (9)$$

$$G_{Voigt} = (1/15)[2c_{11} + c_{33} - (c_{12} + 2c_{23}) + 3(2c_{44} + c_{66})], \quad (10)$$

$$G_{Reuss} = 15[4(2s_{11} + s_{33} - (s_{12} + 2s_{23})) + 3(2s_{44} + s_{66})]^{-1}. \quad (11)$$

Here $s_{ijkl}(=c_{ijkl}^{-1})$ are the elastic compliances (inverse of the elastic constant tensor) and Voigt notation for the compliance tensor is used [Oganov *et al.*, 2002]. Hill averages (Table 2) are estimated as the average of Voigt and Reuss bounds. We compute

$$v_P = \sqrt{\frac{K + \frac{4}{3}G}{\rho}}, \quad v_S = \sqrt{\frac{G}{\rho}}, \quad (12)$$

with K and G the Hill averages of the bulk and shear modulus, respectively.

[29] The anomalies in the single crystal elastic constants (Figure 5) at the loss of magnetism are reflected in the aggregate moduli and sound velocities and, similar to studies in cobalt [Goncharov *et al.*, 2004; Antonangeli *et al.*, 2005; Steinle-Neumann, 2008], should be detectable by high-pressure techniques that can infer acoustic velocities such as inelastic X-ray scattering [Antonangeli *et al.*, 2005; Fiquet *et al.*, 2009] or impulsive stimulated light scattering [Goncharov *et al.*, 2004].

[30] In order to assess our computational results we have compared predictions with ambient NRXIS results at 1 bar and 300 K (Figure 6). The inelastic spectrum contains the information on the partial vibrational density of states (pDOS) of Fe in Fe₇C₃. The pDOS is related to the Debye velocity (v_D) by the following relation:

$$D(E) = \frac{\bar{m}}{2\pi^2 h^3 \rho v_D^3} E^2, \quad (13)$$

Table 2. Elastic Constants c_{ij} (in Voigt Notation), Bulk (K) and Shear (G) Moduli for Fe₇C₃ at Zero Pressure Volume and Their Initial Pressure Derivative^a

| [GPa] | Fe ₇ C ₃ (<i>fm</i>) (This Study) | | Fe ₇ C ₃ (<i>nm</i>) (This Study) | | Fe ₃ C ^b | Fe ₃ C ^c |
|------------|--|--------|--|--------|--------------------------------|--------------------------------|
| | M_0 | M'_0 | M_0 | M'_0 | M_0 | M_0 |
| C_{11} | 397 | 5.7 | 458 | 5.9 | 394 | 417 |
| C_{22} | 397 | 5.7 | 458 | 5.9 | 412 | 416 |
| C_{33} | 247 | 8.6 | 425 | 5.7 | 360 | 381 |
| C_{12} | 173 | 3.1 | 200 | 3.7 | 157 | 157 |
| C_{13} | 168 | 3.9 | 205 | 3.9 | 146 | 171 |
| C_{23} | 168 | 3.9 | 205 | 3.9 | 166 | 174 |
| C_{44} | 102 | 3.1 | 67 | 0.7 | 83 | 82 |
| C_{55} | 102 | 3.1 | 67 | 0.7 | 133 | 136 |
| C_{66} | 112 | 3.3 | 129 | 3.0 | 136 | 140 |
| K_{Hill} | 223 | 4.7 | 301 | 4.4 | 233 | 227 |
| G_{Hill} | 102 | 1.3 | 87 | 1.0 | 115 | 75 |

^aZero pressure volume is represented by subscript zero; its initial pressure derivative is represented by prime.

^bHenriksson *et al.* [2008].

^cJiang *et al.* [2008].

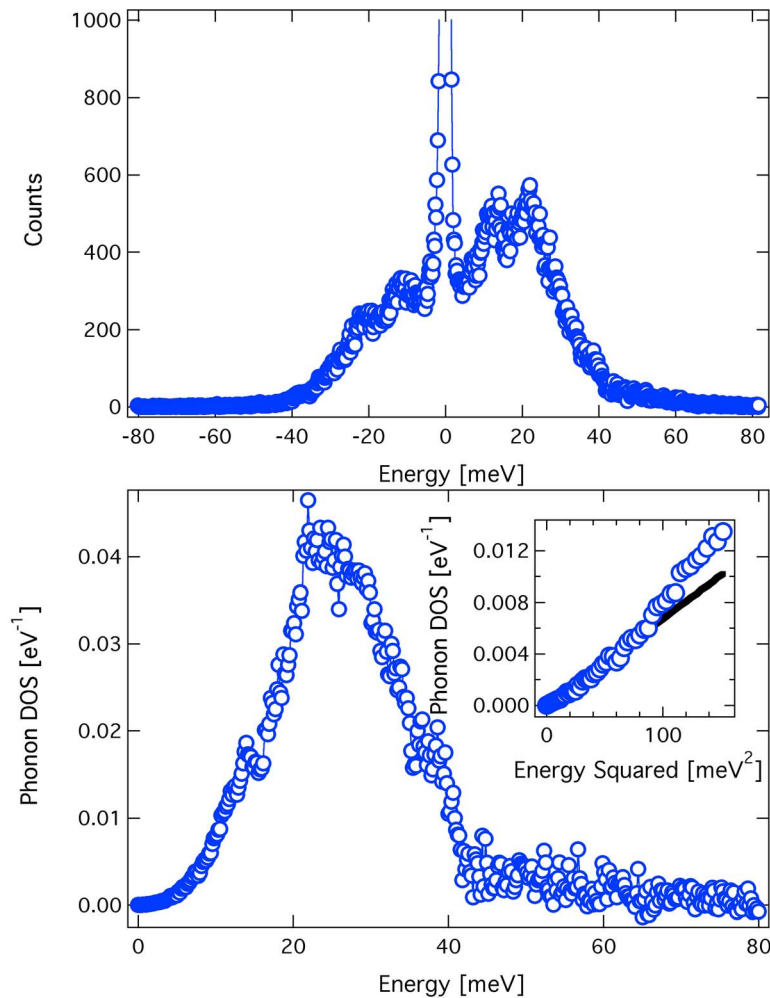


Figure 6. (a) NRXIS spectra of polycrystalline Fe₇C₃ at 1 bar and 300 K and (b) Fe partial phonon density of states pDOS of Fe₇C₃ extracted from NRXIS spectra at 1 bar and 300 K. The inset shows the plot of pDOS versus energy squared for low energies (<12 meV), exhibiting a linear relationship at energies between 0 and 100 meV². The linear relation is used to estimate the Debye velocity (v_D).

where ρ is the density of the material, and \bar{m} is the mass of the nuclear resonant isotope [Sturhahn and Jackson, 2007]. The Debye velocity (v_D) is related to longitudinal (v_P) and shear wave velocity (v_S) by

$$\frac{3}{v_D^3} = \frac{1}{v_P^3} + \frac{2}{v_S^3}. \quad (14)$$

The experimentally determined density of polycrystalline Fe₇C₃ at ambient conditions is around 7.73 g/cm³, and the corresponding v_D is 3.18 km/s. At the same density, the computed velocity is $v_D = 3.97$ km/s, ~20% greater than in experiments. This discrepancy could be associated with the thermal effects on the elastic constants and with the choice of the exchange and correlation potential. It is also important to note that small amounts of impurities in the synthesized samples could severely influence the observed v_D [Sturhahn and Jackson, 2007], similar to

estimates of the magnetic moments [Tajima and Hirano, 1990].

4. Geophysical Significance

[31] In order to constrain the amount of iron carbide that could be present in the inner core, we need to evaluate its density at temperatures relevant for the inner core and combine the results with that of pure iron and then compare values to seismic reference models such as the preliminary reference Earth model (PREM) [Dziewonski and Anderson, 1981].

[32] The temperature of the Earth's core continues to be a subject of current research. Computations on the melting of iron and melting point depression due to the light element present in the core [Alfè *et al.*, 2002a], as well as relating high-pressure and high-temperature shear elasticity of iron to that of the inner core [Steinle-Neumann *et al.*, 2001], point to a temperature of ~5600–5700 K at the inner core

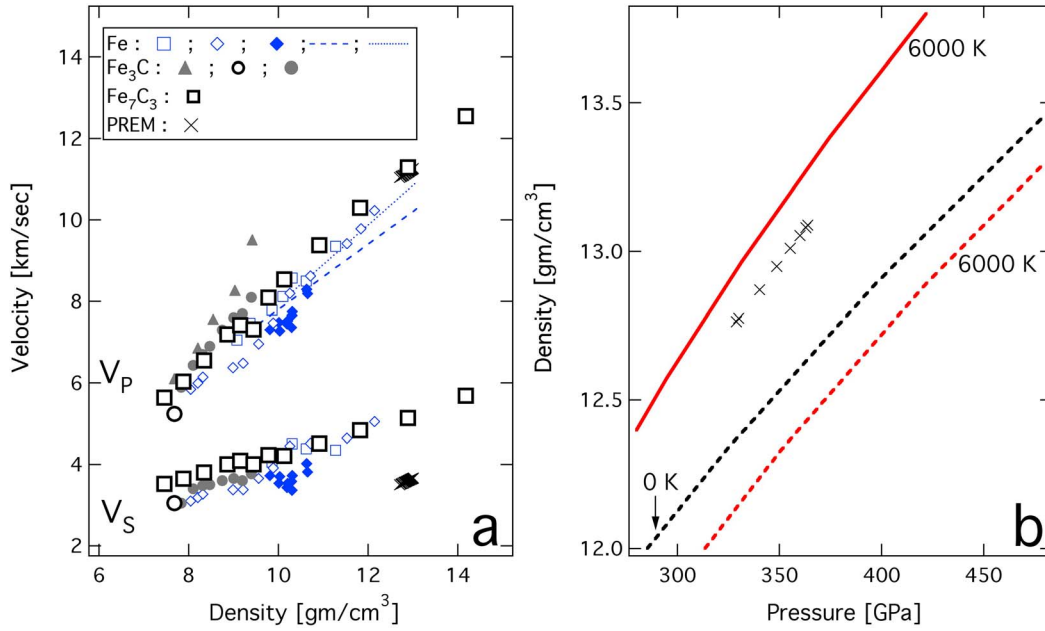


Figure 7. (a) Velocity–density systematics for iron and iron carbide alloys. Compressional wave velocity (v_P) and shear wave velocity (v_S) are plotted with respect to density for the ferromagnetic Fe₇C₃ phase (present study, black open squares). Note the softening of v_P and v_S at the transition from ferromagnetic to nonmagnetic phase at 67 GPa. Also shown are data for Fe₃C by *Gao et al.* [2008] with gray filled circles and by *Dodd et al.* [2003] in black open circles. Data on hcp-Fe at various temperatures are shown in blue symbols: [*Lin et al.*, 2005] (700–1700 K) with filled rhombs, [*Mao et al.*, 2001] (300 K) with open rhombs, [*Fiquet et al.*, 2001] (298 K) with dashed lines, [*Antonangeli et al.*, 2004] (300 K) with a blue open square, and along the Hugoniot [*Brown and McQueen*, 1986] with a dotted line. Large crosses show seismological values for the inner core from preliminary reference Earth model (PREM) [*Dziewonski and Anderson*, 1981]. (b) Densities of Fe₇C₃ (static black dashed line and 6000 K red dashed line) and hcp-Fe [*Isaak and Anderson*, 2003], (6000 K, solid red line). Large crosses show seismological values for the inner core from PREM [*Dziewonski and Anderson*, 1981].

boundary (ICB, 329 GPa). Experimental extrapolations of Fe melting curves to the ICB [*Brown and McQueen*, 1986; *Ma et al.*, 2004; *Nguyen and Holmes*, 2004], however, yield a temperature that is 500–1000 K lower.

[33] We extrapolate the computed density of Fe₇C₃ to the temperature range that is likely to cover inner core conditions with a Debye–Grüneisen model following *Mookherjee and Stixrude* [2009]. The zero-point pressure is given by

$$P_{zp} = \frac{9n\gamma k_B \theta_D}{8V}, \quad (15)$$

where θ_D is the Debye temperature, γ is the Grüneisen parameter, V is the volume of the unit cell, n is the number of atoms in the unit cell, and k_B is the Boltzmann constant. In the Debye approximation, the thermal energy is given by

$$E_{th} = 9nk_B \left(\frac{T}{\theta_D}\right)^3 T \int_0^{\theta_D/T} (x^3/(e^x - 1)) dx \quad (16)$$

and the thermal pressure is

$$P_{th} = \frac{\gamma E_{th}}{V}. \quad (17)$$

The volume dependence of γ and θ_D are approximated with the relations

$$\gamma(V) = \gamma_0 \left(\frac{V}{V_0}\right)^q \quad (18)$$

$$\theta_D = \theta_0 \exp\left(\frac{\gamma_0 - \gamma(V)}{q}\right). \quad (19)$$

We determine $\theta_D \sim 920$ K, $\gamma_0 \sim 2.57$, and $q \sim 2.2$ on the basis of experimental data (*Nakajima et al.*, submitted manuscript, 2010).

[34] The computed static P wave velocity agrees well with the reference Earth model while the calculated S wave velocity is considerably faster (by $\sim 30\%$) than PREM (Figure 7). In comparison, the P and S wave velocities of Fe₃C, extrapolated to inner core conditions, are 14% and 25% greater, respectively [*Gao et al.*, 2008]. It is worth

noting that temperature is likely to reduce the seismic wave velocity by a few percent [Lin *et al.*, 2005], making the agreement of the *P* wave velocity worse and reducing the discrepancy between the *S* wave velocity and PREM [Dziewonski and Anderson, 1981]. Similarly, the difference of v_S for pure iron at high temperatures and the reference Earth model is pronounced, and a number of possible causes for the discrepancy between material properties and seismically observed shear wave velocity have been discussed. It could be related to nonlinear thermoelastic effects at core condition [Steinle-Neumann *et al.*, 2001], attenuation in the inner core with quality factor (*Q*) ~ 200 – 400 [Souriau and Roudil, 1995], or partial melting of around 3–10% [Singh *et al.*, 2000].

[35] As the presence of carbides has been predicted to be stable over the substitutional incorporation of C in the Fe lattice [Huang *et al.*, 2005], Fe₇C₃ is a plausible carbide phase for the inner core coexisting with Fe. As its extrapolated density is lower than that of the inner core [Dziewonski and Anderson, 1981; Kennett *et al.*, 1995], it can account for some of the density deficit (Figure 6). We determine the maximum volume fraction of Fe₇C₃ in the inner core at 6000 K and ICB pressure using the relation

$$\rho_{PREM} = x\rho_{Fe_7C_3} + (1-x)\rho_{Fe}. \quad (20)$$

With $\rho_{Fe} \sim 12.9 \text{ g cm}^{-3}$, $\rho_{Fe_7C_3} \sim 12.1 \text{ g cm}^{-3}$, and $\rho_{PREM} \sim 12.76 \text{ g cm}^{-3}$, we find $x = 18 \text{ vol } \%$ or a maximum possible carbon content of $\sim 1.5 \text{ wt } \%$ in the inner core. This upper limit is a first-order approximation as the inner core might contain other volatiles as well [Jephcoat and Olson, 1987; Stixrude *et al.*, 1997], and the estimated carbon budget of the core varies between 4 wt % [Hillgren *et al.*, 2000] and 0.2 wt % [McDonough, 2003]. Also, the thermoelastic parameters that enter the thermal corrections are not well constrained at inner core conditions.

5. Conclusion

[36] In Fe₇C₃, magnetic collapse occurs at around 67 GPa and is associated with elastic softening in c_{11} , c_{12} , c_{44} . The magnetic collapse is correlated with changes in c/a ratio and reduction of the CFe₆-polyhedral volumes. The calculated *P* wave velocity is in good agreement with seismic models for the Earth's inner core, whereas the *S* wave velocity is larger by 30%. The predicted Debye velocity is also in good agreement with the experimentally determined Debye velocity using NRXIS at ambient conditions. The discrepancy in *S* wave velocity might be related to temperature-induced anelasticity or the presence of partial melts in the inner core. On the basis of the equation of state parameters for the nonmagnetic phase, which is predicted to be stable at pressure and temperature relevant for the inner core, the maximum carbon content in the inner core is likely not to exceed 1.5 wt %.

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A. Chumakov, ID18 Nuclear Resonance Group, European Synchrotron Radiation Facility, Grenoble Cedex F-38043, France.

L. Dubrovinsky, K. Glazyrin, C. McCammon, M. Mookherjee, Y. Nakajima, and G. Steinle-Neumann, Bayerisches Geoinstitut, Universität Bayreuth, Bayreuth D-95440, Germany. (mainak.mookherjee@uni-bayreuth.de)

X. Wu, School of Earth and Space Sciences, Peking University, Beijing 100781, China.